

The IR and UV spectra and an analysis of the conversion products, together with the absence of a depression of the melting point with an authentic sample, enabled glycoside I to be identified as kaempferol 3-O- α -L-rhamnofuranoside-7-O- α -L-rhamnofuranoside, which has been reported previously under the names "kaempferitrin" and "lespedin" [2, 3].

28 March 1969

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UDC 547.972

LUTEOLIN FROM THE LEAVES OF DIGITALIS CILIATA

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Khimiya Prirodnikh Soedinenii, Vol. 5, No. 5, p. 440, 1969

In the preparation of cardiac glycosides from the leaves of Digitalis ciliata Trautv. we isolated a yellow crystalline substance [1] giving all the reactions for flavonoids. In the cyanidin test, an octanol-extractable orange-red pigment was formed, which shows the aglycone nature of the compound [2]. When the substance was subjected to paper chromatography in the butanol-acetic acid-water (4:1:5) system it gave a single spot, while in each of the systems ethyl acetate-formic acid-water (10:2:3) and benzene-ethyl acetate-acetic acid (74.5:23.5:2) systems it gave two spots. On them the main component appeared at the level of an authentic sample of luteolin, and a small spot in the region of apigenin.

To separate the combined flavonoids into the individual compounds we chromatographed them on a polyamide sorbent. Pure luteolin was isolated by washing the column with a mixture of chloroform and ethanol (1:1). After its recrystallization from dilute ethanol, long yellow acicular crystals, $C_{15}H_{10}O_6$, with mp 330-332° C were obtained. It gave no depression with standard luteolin. The acetate of the substance melted at 226-231° C. The IR and UV spectra of the flavonoid and its acetate coincided completely with literature data for luteolin and its acetate [3, 4].

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23 April 1969

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UDC 547.972

LUTEOLIN 7-GLUCOSIDE FROM CAMPANULA LACTIFLORA

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Khimiya Prirodnikh Soedinenii, Vol. 5, No. 5, pp. 440-441, 1969

The chromatography of an ethanolic-aqueous extract from the leaves of Campanula lactiflora M. B. in the BAW (4:1:5) system showed the presence of three substances of a flavonoid nature. One of them, with the composition $C_{20}H_{20}O_{11}$, mp 256-258° C (from ethanol), $[\alpha]_D^{20} -58^\circ$ (c 0.528; methanol-pyridine (3:2)), mol. wt. 259, is a flavone glycoside as was shown by the results of color reactions.

UV spectrum: λ_{\max} 352, 255 m μ ; $\lambda_{\max}^{AlCl_3}$ 400, 275 m μ ; $\lambda_{\max}^{CH_3COONa}$ 355, 258; $\lambda_{\max}^{CH_3COONa+H_3BO_3}$ 380, 258 m μ ; $\lambda_{\max}^{CH_3ONa}$ 407, 265 m μ ; $\lambda_{\max}^{AlCl_3+HCl}$ 390, 275 m μ . The elementary composition found corresponds to that calculated.